

The Atomic Theory as Applied To Gases, with Some Experiments on the Viscosity of Air

by

Silas W. Holman

and

Luisa Hernández

B.S. Mechanical Engineering, UCLA, 2018.

M.S. Stellar Interiors, Vulcan Science Academy, 2020.

M.B.A. Ferengi School of Management, 2022.

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Authored by: Silas W. Holman
Department of Physics
May 18, 1876

Authored by: Luisa Hernández
Department of Physics
May 18, 1876

Certified by: Edward C. Pickering
Professor of Physics, Thesis Supervisor

Accepted by: Quintus Castor
Professor of Wetlands Engineering
Undergraduate Officer, Department of Physics

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ABSTRACT

The developments of the “kinetic theory” of gases made within the last ten years have enabled it to account satisfactorily for many of the laws of gases. The mathematical deductions of Clausius, Maxwell and others, based upon the hypothesis of a gas composed of molecules acting upon each other at impact like perfectly elastic spheres, have furnished expressions for the laws of its elasticity, viscosity, conductivity for heat, diffusive power and other properties. For some of these laws we have experimental data of value in testing the validity of these deductions and assumptions. Next to the elasticity, perhaps the phenomena of the viscosity of gases are best adapted to investigation.¹

Thesis supervisor: Edward C. Pickering
Title: Professor of Physics

¹Text from Holman (1876): doi:[10.2307/25138434](https://doi.org/10.2307/25138434).